

عنوان مقاله:

Catalytic Properties of Ag@Zn-MOF Nanocomposites for Dehydrogenation of Ammonia Borane

محل انتشار:

فصلنامه شیمی معدنی، دوره 5، شماره 1 (سال: 1400)

تعداد صفحات اصل مقاله: 8

نویسندگان:

Reza Sacourbaravi - Department of Chemistry, Faculty of Science, Shahid Chamran University of Ahvaz, Ahvaz, Iran

Zeinab Ansari-Asl - Department of Chemistry, Faculty of Science, Shahid Chamran University of Ahvaz, Ahvaz, Iran

Mohammad Kooti - Department of Chemistry, Faculty of Science, Shahid Chamran University of Ahvaz, Ahvaz, Iran

Valiollah Nobakht - Department of Chemistry, Faculty of Science, Shahid Chamran University of Ahvaz, Ahvaz, Iran

خلاصه مقاله:

The utilization of NH_3BH_3 (ammonium borane) as a H_2 gas storage compound is restricted by its slow rate for H_2 evolution. In this study, three Ag@Zn-MOF nanocomposites with different amounts of Ag:Zn-MOF ratio of 0.25:1 (1), 0.5:1 (2), and 1:1 (3) were investigated as catalysts for hydrogen evolution from hydrolysis of NH_3BH_3 . Well dispersed encapsulated Ag nanoparticles (30-60 nm) in the matrix of the composites have been prepared in the presence of Zn(II) metal-organic frameworks (Zn-MOFs) in an aqueous solution by using NaBH_4 as a reducing agent at room temperature. These nanocomposites have shown good catalytic activity for the hydrolysis of NH_3BH_3 .

کلمات کلیدی:

Hydrogen generation, Ammonia borane, Metal-organic framework, nanocatalyst

لینک ثابت مقاله در پایگاه سیویلیکا:

<https://civilica.com/doc/1189530>

