

عنوان مقاله:

Effects of Substituent and Temperature on the Electronic Properties and Thermodynamics Parameters of 1-(Benzothiazolylamino) Methyl-Y-Naphthol: A Computational Study

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خلاصه مقاله:

The effects of substituent and temperature on the 1- (benzothiazolylamino) methyl - Y- naphthol molecules were studied at CAM-B٣LYP/۶-٣\\G (d,p) level of theory. Selected substitutions were NHY, OH, Me, F, Cl, CHO, COOH, CN and NOY. Substituent effect on the frontier orbitals energies, HOMO-LUMO gap, chemical potential, and electrophilicity were explored. Dependencies of energy changes, reactivity properties and thermodynamics parameters (G, H and S) on the Hammett's constants (p) were provided. Thermodynamics parameters values of the formation reaction were studied reaction at 100-1000 K range. Computational investigation of the substituents and temperature influences on the formation reaction of the 1-(benzothiazolylamino) methyl-Y-naphthol showed the negative E values in the studied reactions. Calculated dipole moment values reveal the presence of the EWGs induced a larger dipole moment compared with the EDGs. Higher negative values were found in the presence of EWGs compared to EDGs. Frontier orbitals were stabilized in the presence of EWGs. However, frontier orbitals were destabilized in the presence of EDGs. Thermodynamic analysis revealed that these reactions were non-spontaneous and exothermic. The G and negative H values were increased with increasing the temperature. Optimization and vibrational analysis were done with Gaussian •9 software package. The standard ۶-۳۱۱G(d,p) basis set was considered for the elements, respectively. CAM-B۳LYP functional was used for the optimization of the geometries of the compound. This functional is Handy and coworkers' long range corrected version of BTLYP using the Coulombattenuating method The identities of the optimized structures as an energy minimum were confirmed by vibrational analysis

كلمات كليدى:

Benzothiazolylamino)methyl-Y-naphthol, Substituent effect, Temperature effect, Thermodynamics parameters)-1

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