

## عنوان مقاله:

Interaction of Vitamin B<sub>3</sub> with Parent Uracil and Anticancer Uracils: A Detailed Computational Approach

## محل انتشار:

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## خلاصه مقاله:

A detailed study on the formed complexes from interaction between vitamin B<sub>3</sub> with parent uracil and anticancer uracil's is performed using M<sub>06-2X</sub>/6-311++G(d,p) and B<sub>3</sub>LYP/6-311++G(d,p) levels of theory. In the studied systems, the uracil's can be placed in three preferential interaction sites (A<sub>1</sub>-A<sub>3</sub>) in the vicinity of the vitamin B<sub>3</sub>. For each uracil group, three configurations corresponding to energetic local minima are obtained. Among the various hydrogen bonding sites, the A<sub>1</sub> region of uracil's shows the strongest interactions at both levels of theory. The analyzed dimers are also stabilized by two hydrogen bonds (H-bonds). The predicted H-bonds in the formation of complexes are: O...H-N and O(S)...H-O. The topological properties of the electron density distribution are also analyzed in terms of the Quantum Theory of "Atoms in Molecules" (QTAIM). Furthermore, the natural bond orbital (NBO) analysis is applied to get a more precise insight about the nature of the H-bond interactions. The calculations reveal that, in most cases, the O(S)...H-O H-bonds are stronger than the O...H-N ones. The calculated energies of highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) show that charge transfer occurs within the molecules.

## کلمات کلیدی:

Vitamin B<sub>3</sub>, Anticancer uracils, DFT, NBO, QTAIM

## لینک ثابت مقاله در پایگاه سیویلیکا:

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