

عنوان مقاله:

Direct Blue YI Removal from Aqueous Solutions by Adsorption on Pistachio Hull Waste: Equilibrium, Kinetic and Thermodynamic Studies

محل انتشار:

مجله علوم پزشکی ایران, دوره 4, شماره 2 (سال: 1395)

تعداد صفحات اصل مقاله: 16

نویسندگان:

Hamed Biglari - Gonabad University of Medical Sciences, Gonabad, Iran

Nasibeh Javan - Zahedan University of Medical Sciences, Zahedan, Iran

Rasoul Khosravi - Birjand University of Medical Sciences, Birjand, Iran

Amin Zarei - Torbateheidarie University of Medical Sciences, Mashhad, Iran

خلاصه مقاله:

Background and Purpose: Azo dyes including Direct Blue YI (DBYI) are toxic, mutagenic and carcinogenic contaminants in effluents of industries. This study aimed to investigate the adsorption of DBY1 from aqueous solution onto pistachio hull waste (PHW) as a low-cost adsorbent. Materials and Methods: A series of experiments were performed via batch adsorption technique to examine the effect of the process variables such as contact time o-YIo minutes, initial dye concentration 10-100 mg/l, pH Y-1Y, adsorbent 0.0Δ-1 g/l, and the processing temperature of YΔ, Fo, and as C. The concentration changes of DBYI were measured using the colorimetric method by the spectrophotometer TAo ultraviolet/visible at the GAY nm wavelength. Moreover, The PHW was characterized by scanning electron microscopy, Fourier transform infrared spectroscopy, Freundlich and Langmuir isotherm model, PHpzc and Brunauer-Emmett-Teller (BET) surface area analysis. Results: Maximum adsorption capacity was 9.FA mg DBY1 per 1 g adsorbent at pH Y, DBY1 100 mg/l, temperature Δ0° C, and time Y10 minutes. In general, by increasing the adsorbent dosage, time, and processing temperature, the removal efficiency was increased; however, incrementing pH and dye concentration had a reverse effect. Maximum BET specific surface and total pore volume on the adsorbent were 1.0°F mY/g and 0.000°Y cmW/g, respectively. The Freundlich isotherm (RY = 0.991Y) model fits the equilibrium data better than the Langmuir isotherm (RY = o.9oYF) model. The adsorption kinetic was found to be well described by the pseudo-second-order model. Thermodynamic analysis indicated that the adsorption process is a spontaneous and endothermic process. Conclusion: PHW can be used well as a low-cost surface adsorbent in the .treatment of DBYI from aquatic environments

كلمات كليدى:

Direct Blue YI, Adsorption, Pistachio Hull, Aqueous

لینک ثابت مقاله در پایگاه سیویلیکا:

https://civilica.com/doc/1837201



