

عنوان مقاله:

Preparation of MWCNT/Schiff base complex of molybdenum modified electrode and its application in simultaneous detection of ascorbic acid and acetaminophen

محل انتشار:

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خلاصه مقاله:

Multi-walled carbon nanotube (MWCNT) and Cis-Dioxo-(E)-۴-amino-N'-(۳-ethoxy-۲-hydroxybenzylidene) benzohydrazide Mo(VI) was used in the modification of carbon-paste electrode (CPE). The modified electrode is used as a sensitive voltammetric sensor for determination of acetaminophen (AC) and ascorbic acid (AA). The electrode showed efficient electrocatalytic activity in lowering the anodic overpotentials and enhancement of the anodic currents. This electrode showed to be able to completely resolve the voltammetric response of AC and AA. The effects of potential sweep rate and pH of the buffer solution on the response of the electrode, toward AC and AA, and the peak resolution is thoroughly investigated by cyclic and differential pulse voltammetry (CV and DPV). The best peak resolution for these compounds using the modified electrode is obtained in solutions with pH = ۴.۰. The ΔE_p for AC and AA in these methods is resulted as about ۳۴۰ mV, which is considerably better than previous reports for these compounds. A linear dynamic range of ۴×10^{-7} to ۱×10^{-3} M and ۲×10^{-6} to ۱×10^{-3} M for AC and AA, respectively in buffered solutions with pH ۴.۰. The detection limit of ۲×10^{-7} M and ۱×10^{-6} is resulted for AC and AA, respectively. The voltammetric detection system was very stable and thereproducibility of the electrode response, based on the six measurements during one month, was less than ۳.۵% for the slope of the calibration curves of AC. The prepared modified electrode is successfully applied for the determination of AA and AC in mixture samples and reasonable accuracies are resulted.

کلمات کلیدی:

Schiff-Base complex of molybdenum; acetaminophen; carbon nanotubes; voltammetry

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