## عنوان مقاله:

Kinetics studies removal of copper ions from aqueous solution using magnetic nanoparticles immobilized on Activated

Carbon

## محل انتشار:

هجدهمین همایش شیمی فیزیک ایران (سال: 1394)

تعداد صفحات اصل مقاله: 2

## نویسندگان:

A Khosravan - Department of New Materials, Institute of Science and High Technology and Environmental Science, Graduate University of Advanced Technology, Kerman, Iran

M.R Sohrabi - Department of Chemistry, North Tehran Branch, Islamic Azad University, Tehran, Iran

M Khosravi - Department of Chemistry, North Tehran Branch, Islamic Azad University, Tehran, Iran

M Davallo - Department of Chemistry, North Tehran Branch, Islamic Azad University, Tehran, Iran

## خلاصه مقاله:

Heavy metals (Copper, Cadmium, Nickel, Lead and Chromium) are natural constituents of the Earth, usually are toxic. These metals, even at trace level, are known to be a risk forhuman beings [1]. The presence of copper, cadmium, nickel and others metals in the aqueousenvironment has a potentially damaging effect on human physiology and other biological systems when the acceptable levels are exceeded [2]. In this study, magnetic nanoparticles(MNP) have been synthesized by co-precipitation method. Fe3O4 immobilized on activated carbon (AC). The removal of copper ions from aqueous solution by synthesized magneticactivated Carbon nanoparticles was investigated.

Adsorption kinetics and isotherm modelswere found

كلمات كليدى:

لینک ثابت مقاله در پایگاه سیویلیکا:

https://civilica.com/doc/552691

