

عنوان مقاله:

Synthesis and Characterization of Fe₃O₄@C-SO₃H as a highly efficient Nano-Catalyst for the preparation of 2/-aminobenzothiazolomethylnaphthols and 1-amidoalkyl-2-naphthols

محل انتشار:

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خلاصه مقاله:

The enhanced activity of heterogeneous (Fe₃O₄@C-SO₃H) ascribed to the high stability of its acid sites, high density, carbon sheets hydrophobic property, and existence of -SO₃H and -COOH groups in its molecular structure [1-5]. A stable core-shell structured magnetic solid acid catalyst Fe₃O₄@C-SO₃H, was prepared from Starch, concentrated sulfuric acid and modified magnetic particles of Fe₃O₄, which was used as the core. The effects of the carbonization and sulfonation processes on the activity of the catalysts were investigated. The result showed that preparation conditions had great influence on the quantity of the acidic groups (sulfonic, carboxyl, and hydroxy groups) and the stability of magnetic catalysts. Then we used the magnetically Fe₃O₄@Sta-SO₃H catalyst as a heterogeneous catalyst for the preparation of 2/-aminobenzothiazolomethylnaphthols and 1- amidoalkyl-2-naphthols via the one-pot three-component reaction of aldehyde, β naphthol and 2-aminobenzothiazole, acetamide or benzamide (Scheme). To our delight, the reaction was efficiently catalyzed by 0.03 g of Fe₃O₄@Sta-SO₃H at 80 °C under Solvent-free conditions, to give full conversion and high yield. It is clear from the result that the aromatic aldehydes bearing electron-withdrawing and electron-donating groups performed equally well in this reaction and give excellent yield of the products within 2–8 h.

کلمات کلیدی:

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